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Acids that contain one ionizable hydrogen (ex  $\text{HNO}_3(\text{aq})$ ,  $\text{HCl}$

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(aq)) are called monoprotic acids If an acid contains two and three ionizable hydrogens (ex  $\text{H}_2\text{SO}_4(\text{aq})$ ,  $\text{H}_3\text{PO}_4(\text{aq})$ ) they are

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bases come in pairs.

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diprotic and triprotic acids, respectively Not all compounds that contain hydrogen are acids (ex  $\text{CH}_4(\text{g})$ ) Also, not all hydrogens of an acid Acids And Bases Chapter 19 Answers Acids And Bases ...

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