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Investigate how calorimetry can be used to find relative specific heat values when different substances are mixed with water. A dynamic graph (temperature vs. time) shows temperatures of the individual substances after mixing.

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Gizmo 24 Worksheets - Lesson Worksheets

Student Exploration- Calorimetry Lab (ANSWER KEY) Gizmo Warm-up A calorimeter is an insulated container filled with a liquid, usually water When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water

[Book] Answer Key For Calorimetry Lab

the initial temperature of 20 °C to result in a final system temperature of 20.45 °C. (Hint: Start by calculating how much heat energy is needed to change the water's temperature by 0.45 °C). Show your work. Use the Gizmo to check your answer. 3000gx 4.184 j/g x 0.45 c = 5,648j x.795 j/g x-59.55C = - 5,64

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