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reduction; (b)

oxidation; (c)

oxidation; (d)

reduction. 3. (a) $F_2 +$

$Ca \rightarrow 2F^- + Ca^{2+}$; (b)

$Cl_2 + 2Li \rightarrow 2Li^+ + 2Cl^-$;

(c) $3Br_2 + 2Fe \rightarrow 2Fe^{3+}$

$+ 6Br^-$; (d) $MnO_4^- +$

$4H^+ + 3Ag \rightarrow 3Ag^+ +$

$MnO_2 + 2H_2O$. 5.

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The instantaneous rate is the rate of a reaction at any particular point in time, a period of time that is so short that the concentrations of reactants and products change by a negligible amount.

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weak.... stress shift
change in
concentration. A
substance that, when
added to a solution,
makes changing the....
weak acid or base with
the salt of it ... or...
make conjugate
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