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Answers

# Chapter 8 Covalent Bonding Assessment Answers

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## **Chapter 8 Covalent Bonding Assessment**

Section 8.1 Assessment page 247 7.

Identify the type of atom that generally forms covalent bonds. The majority of

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covalent bonds form between nonmetallic elements. 8. Describe how the octet rule applies to covalent bonds. Atoms share valence electrons; the shared electrons complete the octet of each atom. 9. Illustrate the formation of single, double, and

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## **Weebly**

Chemistry (12th Edition) answers to Chapter 8 - Covalent Bonding - 8 Assessment - Page 256 56 including work step by step written by community members like you. Textbook Authors: Wilbraham, ISBN-10: 0132525763, ISBN-13: 978-0-13252-576-3, Publisher: Prentice Hall

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## **Chapter 8 - Covalent Bonding - 8 Assessment - GradeSaver**

Chapter 8 - Covalent Bonding - 8  
Assessment - Page 257: 66 Answer

Hydrogen bonds are formed by a  
negative and positive attraction between  
a slightly positive hydrogen atom that is  
covalently bonded to a highly



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electronegative atom (such as Oxygen) and an unshared electron pair of a nearby atom.

## **Chapter 8 - Covalent Bonding - 8 Assessment - GradeSaver**

Chapter 8 - Covalent Bonding - 8

Assessment - Page 257: 68 Answer

Boiling points require the molecules to

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separate , when the state changes from a liquid into a gas.

## **Chapter 8 - Covalent Bonding - 8 Assessment - GradeSaver**

Chapter 8 Covalent Bonding. covalent bond. molecule. diatomic molecule.

BrINClHOF. a bond formed by the sharing of electrons between atoms. a

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neutral group of atoms joined together by covalent bonds. a molecule consisting of two atoms of the same element. the mnemonic for remembering which non-metal elements form dia....

## **chapter 8 test chemistry covalent bonding Flashcards and ...**

Section 8.2 - The Nature of Covalent

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Bonding In ionic bonding, atoms transfer electrons to achieve noble gas configuration. In covalent bonding, atoms share electrons to achieve noble gas configuration. Most atoms share electrons until they have a total of 8 valence electrons (octet rule).

## **Chapter 8 - Covalent Bonding**

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## Covalent Bonding Assessment

### Answers

242 Chapter 8 • Covalent Bonding Single Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond. The shared electron pair is often referred to as the bonding pair. For a hydrogen molecule, shown in Figure 8.4, each covalently bonded atom equally attracts the pair of shared

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electrons.

## **Chemistry Chapter 8 Covalent Bonding Assessment Answer Key**

1. In covalent bonds, electron sharing usually occurs so that atoms attain the electron configuration of noble gases.
- (8) 2. Atoms form double or triple covalent bonds if they can attain a noble

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gas structure by sharing two pairs or three pairs of electrons.

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Chapter 8 Covalent Bonding Study Guide: McGraw Hill Textbook. Flashcard maker : Lily Taylor. When sharing of electrons occurs the attachment

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between atoms is called. covalent bond.  
in a covalent bond, the dissociation  
energy is released in the process of.  
exothermic reaction.

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240 Chapter 8 • Covalent Bonding  
Section 88.1.1 The Covalent



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Bond-!).)DEAAtoms gain stability when they share electrons and form covalent bonds. Real-World Reading Link Have you ever run in a three-legged race? Each person in the race shares one of their legs with a teammate to form a single three-legged team.

## **Chapter 8: Covalent Bonding**

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a covalent bond between two atoms in which the shared electron pair comes from only one of the atoms. ... Chapter 8 Covalent Bonding Part B Test Review. 104 terms. lilylover123. Chemistry Chapter 8 Covalent Bonding. 31 terms. adamjgillman. Chemistry Chapter 8 Covalent Bonding. 57 terms. Blubka.

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## Answers

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A chemical bond consisting of a hydrogen atom between two electronegative atoms (e.g., oxygen or nitrogen) with one side being a covalent bond and the other being an ionic bond; the attractive force between the hydrogen attached to an electronegative atom of one molecule and an

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electronegative atom of a different molecule.

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Worksheet Lewis Diagrams of Covalent Compounds Complete the worksheet. Be sure to show your work. Chapter 8 Section 8.2 The Nature of Covalent Bonding

## **Term 2, Module 3 (Core) Chapter 8, Covalent Bonding**

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Covalent Bonds When only one pair of electrons is shared, such as in a hydrogen molecule, it is a single covalent bond The shared electron pair is often referred to as the bonding pair For a hydrogen molecule, shown in Figure 84, each covalently bonded atom equally attracts the pair of shared



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