

Chemistry Chapter 12 Stoichiometry

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Acces PDF Chemistry Chapter 12 Stoichiometry Worksheet Quantitative calculations involving reactions in solution are carried out in the same manner as we discussed in Chapter 11.Instead of masses, however, we use volumes of solutions of known concentration to determine the number of moles of reactants.

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Example 1. How many molecules of SO 3 are needed to react with 144 molecules of Fe 2 O 3 given this balanced chemical equation?. Fe 2 O 3 (s) + 3SO 3 (g) → Fe 2 (SO 4) 3. Solution. We use the balanced chemical equation to construct a conversion factor between Fe 2 O 3 and SO 3.The number of molecules of Fe 2 O 3 goes on the bottom of our conversion factor so it cancels with our given amount ...