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Chemistry Chapter 13 review. absolute zero. universal gas constant. Boyle's law. Charles's law. a temperature at which a gas theoretically occupies zero volume. 0.08206 L atm/mol K. at constant temperature, the volume of a given amount of gas i.... at constant pressure, the volume of a given amount of gas is d....

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Chemistry Chapter 13. kinetic-molecular theory. particle size of gases. particle motion of gases. temperature. describes the behavior of gases in terms of particle motion. small, separated by empty space. constant, random motion. a measure of the average kinetic energy of the particles in a....

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CHAPTER 13 REVIEW ANSWER KEY 12. A battery is like a water pump because the battery supplies electrical potential energy to a circuit and a pump provides potential energy to water. 13. Multimeters can measure both voltage and current for AC and DC while ammeters measure only current and voltmeters only voltage (potential difference). 14.

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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Use the guidelines in Table 1on page 437 of the text to predict the solubility of the following compounds in water: soluble a. magnesium nitrate insoluble b. barium sulfate insoluble c. calcium carbonate

13 Ions in Aqueous Solutions and Colligative Properties

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You are given a box containing NH 3, N 2, and H 2 at equilibrium at 1000°C. Analysis of the contents shows that the concentration of NH 3 is 0.102 mole/liter, N 2 is 1.03 moles/liter, and H 2 is 1.62 moles/liter. Calculate K for the reaction: 2NH 3 (g) ↔ N 2 (g) + 3H 2 (g)

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