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Trace Elements. 259. Chapter 7: Trace Elements in Igneous Processes. 7.1 INTRODUCTION. In this chapter we will consider the behavior of trace elements, particularly in magmas, and introduce methods to model this behavior. Though trace elements, by definition, constitute only a small fraction of a system of interest, they provide geochemical and geological information out of proportion to their abundance.

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W. M. White Geochemistry Chapter 1: Introduction. 12A ugst 2 5, 0. (the electropositive element) gives up an electron, becoming positively charged, to the Cl atom (the electronegative element), which becomes negatively charged. Electrostatic forces between the Na^+ and the Cl^- ions hold the ions in place in the crystal.

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William White teaches geochemistry as a Professor of earth and atmospheric sciences at Cornell University. He

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received a B.A. in geology from the University of California, Berkeley and a PhD in oceanography from the University of Rhode Island.

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distinct phase from the tea, but the dissolved sugar is not. Phase is not synonymous with compound. Phases need not be chemically distinct: a glass of ice water has two distinct phases: water and ice. Many solid compounds can exist as more than one phase. Nor

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need they be compositionally

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W. M. White Geochemistry Chapter 1: Introduction 2 to view the same minerals on almost the atomic scale. Techniques such as X-ray diffraction, nuclear magnetic resonance, and Raman and infrared spectroscopy allow us to examine atomic ordering and bonding in natural materials. Mass spectrometers allow us to determine the age of rocks and the tem-

Chapter 1: Introduction

William M. White This book provides a comprehensive introduction to radiogenic and stable isotope geochemistry. Beginning with a brief overview of nuclear physics and nuclear origins, it then reviews radioactive decay schemes and their use in geochronology.

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W. M. White Geochemistry Chapter 8: Radiogenic Isotope Geochemistry 320 January 10, 2001 also binds quarks together to form hadrons, a class of particles that includes neutrons and protons. The intensity of the strong force decreases rapidly with distance, so that at distances more than about 10-14 m it is weaker than the electromagnetic force.

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Geochemistry by William White is an excellent review of the main themes in geochemistry. It is quite complete and goes through the different subjects with very clear and consistent expositions. As a geochemistry professor for more than 30 years I do recommend this book.

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information out of proportion to their
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nuclear origins, it then reviews
radioactive decay schemes and their use
in geochronology. A following chapter
covers the closely related techniques

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such as fission-track and carbon-14 dating.

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Geochemistry 9.2.1.1 The Quantum
Mechanical Origin of Isotopic
Fractionations It is fairly easy to
understand, at a qualitative level at
least, how some isotope fractionations
can arise from vibrational motion.

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Applications of Thermodynamics «' 1999
W. M. White 115 October 4, 1999 or $\ln \frac{W}{G} = -\frac{U}{RT}$
then $\ln h$ then 4.15 is Henry's Law. Thus
the interaction parameter can be related
to the parameters of Henry's Law, and
activity coefficient.

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next three chapters focus on processes at the surface of the Earth. Here water is the dominant substance, and the tools of thermodynamics, kinetics, and aquatic chemistry will be of great use.

W. M. White Geochemistry Chapter 1: Introduction

William White teaches geochemistry as a Professor of earth and atmospheric sciences at Cornell University. He received a B.A. in geology from the University of California, Berkeley and a PhD in oceanography from the University of Rhode Island.

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In the previous 2 chapters, we developed the fundamental thermodynamic relationships and saw how they are applied to geochemical problems. The tools now in our thermodynamic toolbox are sufficient to deal with most of the phenomena we will encounter in the

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second half of this book.

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